

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

**catena-Poly[[[2-(pyridin-2-yl)disulfanyl]-pyridine- $\kappa^2$ N,S]copper(I)]- $\mu_{1,5}$ -dicyanamido]**

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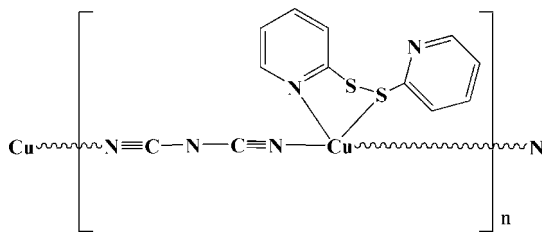
Received 8 January 2011; accepted 20 January 2011

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.028;  $wR$  factor = 0.081; data-to-parameter ratio = 14.7.

In the title compound,  $[\text{Cu}(\text{C}_2\text{N}_3)(\text{C}_{10}\text{H}_8\text{N}_2\text{S}_2)]_n$ , the  $\text{Cu}^{\text{I}}$  atoms are connected by bridging dicyanamide ligands, forming chains parallel to  $[100]$ . Each  $\text{Cu}^{\text{I}}$  atom displays a tetrahedral coordination environment, formed by one S atom and three N atoms from one 2-(pyridin-2-yl)disulfanylpyridine and two dicyanamide ligands. The crystal structure is stabilized by  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bonds, forming a three-dimensional network.

## Related literature

For potential applications of metal-organic frameworks, see: Eddaoudi *et al.* (2001). For metal-organic frameworks constructed from flexible ligands, see: Xu *et al.* (2009). For related structures, see: Mal *et al.* (2006); Schlueter *et al.* (2007); Sen *et al.* (2007).



## Experimental

## Crystal data

 $[\text{Cu}(\text{C}_2\text{N}_3)(\text{C}_{10}\text{H}_8\text{N}_2\text{S}_2)]$   
 $M_r = 349.92$ 

 Triclinic,  $P\bar{1}$   
 $a = 7.6294$  (15) Å

 $b = 9.5964$  (19) Å  
 $c = 10.202$  (2) Å  
 $\alpha = 84.19$  (3)°  
 $\beta = 80.63$  (3)°  
 $\gamma = 70.93$  (3)°  
 $V = 695.6$  (2) Å<sup>3</sup>
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 1.87$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.20 \times 0.16 \times 0.12$  mm

## Data collection

 Bruker APEXII CCD  
 diffractometer  
 Absorption correction: multi-scan  
 (SADABS; Sheldrick, 1996)  
 $T_{\text{min}} = 0.893$ ,  $T_{\text{max}} = 1.000$ 

 6615 measured reflections  
 2669 independent reflections  
 2301 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.018$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.028$   
 $wR(F^2) = 0.081$   
 $S = 1.07$   
 2669 reflections

 181 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.39$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.29$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C9}-\text{H9}\cdots\text{N3}^{\text{i}}$	0.93	2.53	3.453 (3)	171

Symmetry code: (i)  $x + 1, y - 1, z$ .

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

The authors are grateful for the financial support of the National Natural Science Foundation of China (Nos. 50602024, 50972060) and NUST Research Funding (No. 2010ZDJH06).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZQ2085).

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**supplementary materials**

*Acta Cryst.* (2011). E67, m285 [ doi:10.1107/S1600536811002728 ]

***catena*-Poly[[[2-(pyridin-2-yl)disulfanyl]pyridine- $\kappa^2$ N,S]copper(I)]- $\mu_{1,5}$ -dicyanamido]**

**S. Wu, W. Jiang, F. Li and L. Liu**

**Comment**

Metal-organic compounds have attracted much attention because of their diverse structures (Eddaoudi *et al.*, 2001). Flexible ligands can play different roles in constructing metal-organic frameworks (Xu *et al.*, 2009). The title compound,  $\{C_{12}H_8CuN_5S_2\}_n$ , is constructed by two kinds of flexible ligands: bridging dicyanamide ligands and chelate 2-(pyridin-2-yl)disulfanylpyridine ligands. In this paper, the crystal structure of the title compound is presented. As illustrated in Fig. 1, the Cu atoms are connected by bridging dicyanamide ligands, forming a serrate chain. Each Cu atom displays a tetrahedral coordination environment, formed by one S atom and three N atoms from one 2-(pyridin-2-yl)disulfanylpyridine and two dicyanamide ligands, where the Cu—S and average Cu—N bonds are 2.472 (1) and 1.969 Å, respectively. The crystal structure is stabilized by C—H $\cdots$ N hydrogen bonds [H9 $\cdots$ N3<sup>iii</sup> = 2.53 Å, C9 $\cdots$ N3<sup>iii</sup> = 3.453 (3) Å, and C9—H9 $\cdots$ N3<sup>iii</sup> = 171°] between the central N atom of the dicyanamide and one H atom of a pyridine ring, forming a three-dimensional network [symmetry code: (iii)  $x + 1, y - 1, z$ ].

**Experimental**

CuN(CN)<sub>2</sub> (0.4 mmol) and NaN(CN)<sub>2</sub> (1.2 mmol) were added into 2 ml DMF with thorough stir for 6 minutes. After filtration, the colorless filtrate was carefully laid on the surface with the solution of bis(2-pyridyl)disulfide (0.5 mmol) in 5 ml *i*-PrOH. Colorless crystals were obtained after 3 weeks.

**Refinement**

The H atoms were positioned geometrically and refined with a riding model, with C—H = 0.93 Å and  $U_{iso} = 1.2U_{eq}(C)$ .

**Figures**

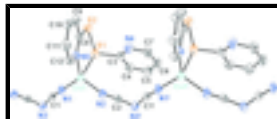


Fig. 1. The molecular structure of a portion of the title compound, with atom labels and 30% probability displacement ellipsoids. All H atoms have been omitted [symmetry code: (i)  $x + 1, y, z$ ].

***catena*-Poly[[[2-(pyridin-2-yl)disulfanyl]pyridine-  $\kappa^2$ N,S]copper(I)]- $\mu_{1,5}$ -dicyanamido]**

*Crystal data*

[Cu(C<sub>2</sub>N<sub>3</sub>)(C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>)]

$M_r = 349.92$

Triclinic,  $P\bar{1}$

$Z = 2$

$F(000) = 352$

$D_x = 1.671 \text{ Mg m}^{-3}$

# supplementary materials

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Hall symbol: -P 1  
 $a = 7.6294 (15) \text{ \AA}$   
 $b = 9.5964 (19) \text{ \AA}$   
 $c = 10.202 (2) \text{ \AA}$   
 $\alpha = 84.19 (3)^\circ$   
 $\beta = 80.63 (3)^\circ$   
 $\gamma = 70.93 (3)^\circ$   
 $V = 695.6 (2) \text{ \AA}^3$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
Cell parameters from 3088 reflections  
 $\theta = 3.0\text{--}28.9^\circ$   
 $\mu = 1.87 \text{ mm}^{-1}$   
 $T = 293 \text{ K}$   
Block, colourless  
 $0.20 \times 0.16 \times 0.12 \text{ mm}$

## Data collection

Bruker APEXII CCD  
diffractometer  
Radiation source: fine-focus sealed tube  
graphite  
 $\omega$  scans  
Absorption correction: multi-scan  
(*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.893$ ,  $T_{\max} = 1.000$   
6615 measured reflections

2669 independent reflections  
2301 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.018$   
 $\theta_{\max} = 26.0^\circ$ ,  $\theta_{\min} = 3.1^\circ$   
 $h = -8 \rightarrow 9$   
 $k = -11 \rightarrow 11$   
 $l = -12 \rightarrow 12$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.028$   
 $wR(F^2) = 0.081$   
 $S = 1.07$   
2669 reflections  
181 parameters  
0 restraints

Primary atom site location: structure-invariant direct methods  
Secondary atom site location: difference Fourier map  
Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0478P)^2 + 0.0204P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.39 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.29 \text{ e \AA}^{-3}$

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cu1	-0.03943 (3)	0.83530 (3)	0.16964 (3)	0.05693 (13)
S1	-0.00326 (7)	0.67390 (7)	0.37486 (5)	0.05609 (17)
S2	0.11161 (9)	0.47767 (7)	0.28919 (5)	0.06255 (18)
N1	-0.3026 (2)	0.9293 (2)	0.16879 (19)	0.0580 (5)
N2	0.1131 (2)	0.9633 (2)	0.17961 (19)	0.0599 (5)
N3	-0.6377 (2)	1.0592 (2)	0.2077 (2)	0.0694 (6)
N4	0.3589 (3)	0.6469 (2)	0.38308 (18)	0.0610 (5)
N5	0.1249 (2)	0.65814 (18)	0.06703 (15)	0.0464 (4)
C1	-0.4605 (3)	0.9840 (2)	0.1839 (2)	0.0489 (5)
C2	0.2361 (3)	1.0016 (2)	0.1905 (2)	0.0480 (5)
C3	0.1841 (3)	0.7106 (2)	0.43562 (18)	0.0452 (4)
C4	0.1311 (3)	0.8103 (3)	0.5330 (2)	0.0612 (6)
H4	0.0059	0.8499	0.5684	0.073*
C5	0.2682 (4)	0.8504 (3)	0.5770 (2)	0.0709 (7)
H5	0.2372	0.9197	0.6414	0.085*
C6	0.4516 (4)	0.7863 (3)	0.5242 (2)	0.0692 (7)
H6	0.5475	0.8109	0.5519	0.083*
C7	0.4889 (3)	0.6858 (3)	0.4302 (2)	0.0695 (7)
H7	0.6136	0.6409	0.3963	0.083*
C8	0.1811 (3)	0.5212 (2)	0.11824 (18)	0.0462 (5)
C9	0.2919 (4)	0.4027 (2)	0.0454 (2)	0.0627 (6)
H9	0.3245	0.3078	0.0850	0.075*
C10	0.3528 (4)	0.4264 (3)	-0.0852 (2)	0.0681 (7)
H10	0.4289	0.3484	-0.1362	0.082*
C11	0.2996 (4)	0.5682 (3)	-0.1403 (2)	0.0660 (6)
H11	0.3402	0.5877	-0.2292	0.079*
C12	0.1868 (3)	0.6795 (2)	-0.0633 (2)	0.0568 (5)
H12	0.1505	0.7748	-0.1019	0.068*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cu1	0.03597 (17)	0.0595 (2)	0.0745 (2)	-0.01273 (13)	-0.00278 (13)	-0.01572 (14)
S1	0.0434 (3)	0.0798 (4)	0.0469 (3)	-0.0244 (3)	0.0007 (2)	-0.0063 (3)
S2	0.0847 (4)	0.0645 (4)	0.0502 (3)	-0.0423 (3)	-0.0082 (3)	0.0048 (3)
N1	0.0393 (10)	0.0623 (11)	0.0748 (12)	-0.0157 (9)	-0.0099 (8)	-0.0125 (9)
N2	0.0394 (10)	0.0572 (11)	0.0836 (13)	-0.0155 (9)	-0.0067 (9)	-0.0086 (10)
N3	0.0382 (10)	0.0433 (10)	0.1278 (17)	-0.0077 (8)	-0.0184 (10)	-0.0154 (11)
N4	0.0464 (10)	0.0763 (13)	0.0606 (10)	-0.0187 (9)	0.0016 (8)	-0.0213 (10)
N5	0.0448 (9)	0.0463 (9)	0.0459 (9)	-0.0120 (7)	-0.0051 (7)	-0.0025 (7)
C1	0.0433 (12)	0.0430 (10)	0.0654 (12)	-0.0170 (9)	-0.0154 (10)	-0.0011 (9)
C2	0.0353 (10)	0.0381 (10)	0.0647 (12)	-0.0040 (8)	-0.0064 (9)	-0.0028 (9)
C3	0.0450 (11)	0.0532 (11)	0.0357 (9)	-0.0147 (9)	-0.0049 (8)	0.0008 (9)
C4	0.0542 (13)	0.0674 (14)	0.0528 (11)	-0.0058 (11)	-0.0028 (10)	-0.0145 (11)

## supplementary materials

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C5	0.0843 (18)	0.0700 (15)	0.0601 (13)	-0.0197 (14)	-0.0148 (12)	-0.0186 (12)
C6	0.0719 (16)	0.0802 (17)	0.0667 (14)	-0.0325 (14)	-0.0237 (13)	-0.0036 (13)
C7	0.0453 (12)	0.0879 (18)	0.0762 (15)	-0.0188 (12)	-0.0066 (11)	-0.0184 (14)
C8	0.0496 (11)	0.0484 (11)	0.0467 (10)	-0.0227 (9)	-0.0093 (9)	-0.0022 (9)
C9	0.0802 (17)	0.0417 (11)	0.0665 (14)	-0.0190 (11)	-0.0118 (12)	-0.0030 (11)
C10	0.0834 (18)	0.0525 (13)	0.0635 (14)	-0.0162 (12)	0.0008 (12)	-0.0167 (11)
C11	0.0817 (17)	0.0652 (15)	0.0466 (11)	-0.0200 (13)	0.0006 (11)	-0.0081 (11)
C12	0.0596 (13)	0.0518 (12)	0.0520 (11)	-0.0108 (10)	-0.0056 (10)	0.0024 (10)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

Cu1—N1	1.9143 (18)	C3—C4	1.369 (3)
Cu1—N2	1.967 (2)	C4—C5	1.377 (4)
Cu1—N5	2.0244 (18)	C4—H4	0.9300
Cu1—S1	2.4720 (10)	C5—C6	1.373 (4)
S1—C3	1.792 (2)	C5—H5	0.9300
S1—S2	2.0207 (11)	C6—C7	1.361 (4)
S2—C8	1.787 (2)	C6—H6	0.9300
N1—C1	1.138 (3)	C7—H7	0.9300
N2—C2	1.138 (3)	C8—C9	1.378 (3)
N3—C2 <sup>i</sup>	1.298 (3)	C9—C10	1.360 (3)
N3—C1	1.303 (3)	C9—H9	0.9300
N4—C3	1.319 (3)	C10—C11	1.375 (3)
N4—C7	1.337 (3)	C10—H10	0.9300
N5—C8	1.322 (3)	C11—C12	1.359 (3)
N5—C12	1.356 (3)	C11—H11	0.9300
C2—N3 <sup>ii</sup>	1.298 (3)	C12—H12	0.9300
N1—Cu1—N2	117.16 (8)	C6—C5—C4	118.9 (2)
N1—Cu1—N5	126.58 (8)	C6—C5—H5	120.6
N2—Cu1—N5	107.60 (7)	C4—C5—H5	120.6
N1—Cu1—S1	106.75 (7)	C7—C6—C5	118.1 (2)
N2—Cu1—S1	104.29 (6)	C7—C6—H6	121.0
N5—Cu1—S1	87.90 (5)	C5—C6—H6	121.0
C3—S1—S2	105.95 (8)	N4—C7—C6	124.6 (2)
C3—S1—Cu1	102.35 (7)	N4—C7—H7	117.7
S2—S1—Cu1	98.20 (4)	C6—C7—H7	117.7
C8—S2—S1	105.52 (8)	N5—C8—C9	123.48 (18)
C1—N1—Cu1	172.06 (19)	N5—C8—S2	121.22 (16)
C2—N2—Cu1	161.63 (17)	C9—C8—S2	115.29 (16)
C2 <sup>i</sup> —N3—C1	120.23 (19)	C10—C9—C8	119.1 (2)
C3—N4—C7	115.9 (2)	C10—C9—H9	120.5
C8—N5—C12	116.52 (18)	C8—C9—H9	120.5
C8—N5—Cu1	124.88 (13)	C9—C10—C11	118.6 (2)
C12—N5—Cu1	118.59 (14)	C9—C10—H10	120.7
N1—C1—N3	173.2 (2)	C11—C10—H10	120.7
N2—C2—N3 <sup>ii</sup>	173.5 (2)	C12—C11—C10	119.2 (2)
N4—C3—C4	124.4 (2)	C12—C11—H11	120.4
N4—C3—S1	119.92 (16)	C10—C11—H11	120.4

C4—C3—S1	115.61 (16)	N5—C12—C11	123.1 (2)
C3—C4—C5	118.1 (2)	N5—C12—H12	118.5
C3—C4—H4	120.9	C11—C12—H12	118.5
C5—C4—H4	120.9		

Symmetry codes: (i)  $x-1, y, z$ ; (ii)  $x+1, y, z$ .

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C9—H9 $\cdots$ N3 <sup>iii</sup>	0.93	2.53	3.453 (3)	171

Symmetry codes: (iii)  $x+1, y-1, z$ .

Fig. 1

